SARANSH | CHEMISTRY



TARGET : NEET (UG) 2024



Course : SARANSH (Youtube Live CRASH COURSE)

DPP No. : 1

1. The correct sequence of reagents for following reaction is :

 $CH_3-CH_3 \longrightarrow CH_3-CH_2-CN$

- (1) (i) HCI (ii) KCN
- (2) KCN
- (3) (i) Cl₂/hv (monochlorination) (ii) KCN
- (4) (i) NaCl (ii) KCN
- 2. The major products formed in the following reaction are :



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3. Given below are two statements:

Statement I: In Lucas test, primary, secondary and tertiary alcohols are distinguished on the basis of their reactivity with conc. HCl + ZnCl₂, known as Lucas Reagent.

Statement II: Primary alcohols are most reactive and immediately produce turbidity at room temperature on reaction with Lucas Reagent.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Statement I is correct but Statement II is incorrect
- (2) Statement I is incorrect but Statement II is correct
- (3) Both Statement I and Statement II are correct
- (4) Both Statement I and Statement II are incorrect
- 4. Which of the following statement is correct for nucleophile ?
 - (1) Electron rich species are called nucleophile.
 - (2) Nucleophiles are Lewis bases.
 - (3) Nucleophile donates long pair of electron to vacant orbital of carbon atom.
 - (4) All are correct.
- 5. S_N1 reactions are favoured by -
 - (1) Non-polar solvents.
 - (2) Bulky groups on the carbon atom attached to the halogen atom.
 - (3) Small groups on carbon atom attached to the halogen atom.
 - (4) None of these.

6. When the concentration of alkyl halide is tripled and the concentration of $\stackrel{\Theta}{OH}$ ion is reduced to half, the rate of S_N2 reaction increases by:

(1) 3 times (2) 2 times	s (3) 1.5 times	(4) 6 times
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- 7. S_N2 mechanism proceeds through intervention of
 - (1) Carbonium ion (2) Transition state (3) Free radical (4) Carbanion



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Answer Key													
1.	(3)	2.	(2)	3.	(1)	4.	(4)	5.	(2)	6.	(3)	7.	(2)
8.	(2)	9.	(3)	10.	(3)								

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