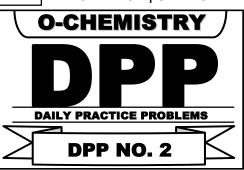
TARGET: NEET (UG) 2024

Course: SARANSH (Youtube Live CRASH COURSE)



DPP No. : 2

1. In the following reaction sequence product Y is

$$C_2H_5Br \xrightarrow{KCN} X \xrightarrow{Dil. H_2SO_4} Y$$

(1) Ethanol

(2) Ethanal

(3) Propanoic acid

(4) Ethanenitrile

2. The reaction: RCOOAg + Br₂ $\xrightarrow{CCl_4$, Reflux \rightarrow R-Br +AgBr + CO₂ is called

(1) Wurtz reaction

(2) Hunsdiecker reaction

(3) Friedel-Crafts reaction

(4) Kolbe's reaction

3. The reaction, $CH_3COOH + CH_3OH \xrightarrow{H^+} CH_3COOCH_3 + H_2O$ is called

(1) Acidification reaction

(2) Dehydration reaction

(3) Dehydrogenation reaction

(4) Esterification reaction

4. The product formed by the reaction of acetamide with Br₂ in presence of NaOH is :

(1) CH₃CN

(2) CH₃CHO

(3) CH₃CH₂OH

(4) CH₃NH₂

5. The decreasing order of reactivity towards nucleophilic acyl substitution is

(i) CH₃COCI

(ii) CH₃COOC₂H₅,

(iii) CH₃CONH₂

(iv) (CH₃CO)₂O

(1) (i) > (iv) > (iii) > (ii)

(2) (i) > (iv) > (ii) > (iii)

(3) (iv) > (iii) > (i) > (ii)

(4) (iii) > (i) > (iv) > (ii)

6. What product is formed when acetic acid reacts with P_2O_5 .

(1) Acetyl chloride

(2) Trichloro acetic acid

(3) Acetic anhydride

(4) Di-chloro acetic acid

7. Process by which formation of acetone take place :

(1) Pyrolysis of CH₃COOH

(2) Oxidation of CH₃COOH

(3) Pyrolysis of calcium acetate

(4) Oxidation of n-propyl alcohol

- **8.** A dihaloalkane on alkaline hydrolysis produces a ketone with formula C₃H₆O. The dihaloalkane can be
 - (1) 2,2-Dichloropropane

(2) 1,1-Dichloropropane

(3) 1,2-Dichloropropane

- (4) 1,3-Dichloropropane
- **9.** The general order of reactivities of given carbonyl compounds towards nucleophilic addition reaction is :
 - (1) $H_2C=O > (CH_3)_2C=O > Ar_2C=O > CH_3CHO > ArCHO$.
 - (2) $H_2C=O > CH_3CHO > (CH_3)_2 C=O > ArCHO > Ar_2C=O$.
 - (3) ArCHO > Ar₂C=O > CH₃CHO > (CH₃)₂ C=O > H₂C=O.
 - (4) $Ar_2C=O > (CH_3)_2C=O > ArCHO > CH_3CHO > H_2C=O$.

10.
$$CHO$$

$$+ HCHO \xrightarrow{KOH} (A) + (B)$$

$$OCH_3$$

(4) (1) and (2) both

Answer Key

- **1.** (3)
 - 3) **2.**
- (2)
- 3.
- (4)
- 5.
- (2)
- 6.
- (3)
- **7.** (3)

- **8.** (1)
- 9.
- (2)
- 10.
- (1)