



TARGET : NEET (UG) 2024

Course : SARANSH (Youtube Live CRASH COURSE)

P/I-CHEMISTRY

**DPP**

DAILY PRACTICE PROBLEMS

**DPP NO. 1**

## Organic Chemistry : Some Basic Principles and Techniques

### DPP No. : 1

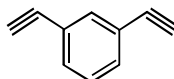
SR. No.	DPPs Qs. Details		Marking Scheme				Time Details		
	Type of Questions	Code	Full Marks	(-ve Marks)	Total Ques.	Total Marks	Qs (in Min.) for Each Qs	Time (in Min.)	Max. Time (in Min.)
1	MULTIPLE CHOICE QUESTION (ONLY ONE CORRECT OPTION)	MCQ	4	-1	10	40	1		10
	<b>Total</b>				<b>10</b>	<b>40</b>			<b>10</b>

1. How many  $sp^2-sp$   $\sigma$ -bonds found in the given compound.



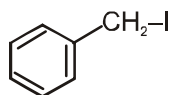
- (1) 2                                      (2) 3                                      (3) 4                                      (4) 5

2. How many C-C sigma ( $\sigma$ ) bonds are present in the given compound?



- (1) 8                                      (2) 9                                      (3) 10                                      (4) 11

3. Which of the following statement is / are correct for following compound




- (1) It is an allylic iodide                                      (2) It is a vinylic iodide  
 (3) It is 2° benzylic iodide                                      (4) It is 1° benzylic iodide

4. Which of the following is an allyl group

- (1)  $CH_2 = CH -$                                       (2)  $CH_2 = CH - CH_2 -$   
 (3)  $CH_2 = CH - CH_2 - CH_2 -$                                       (4)  $CH_3 - CH = CH -$



5.  $\text{CH}_2 = \text{CH} - \text{Br}$  is a / an  
 (1) Allylic bromide      (2) Benzylic bromide      (3) Vinylic bromide      (4) All of these
6. How many  $\pi$  and  $\sigma$  bonds are in the given compound ?  
 $\text{HC} \equiv \text{C} - \text{CH} = \text{CH}_2$   
 (1)  $3\pi$  and  $6\sigma$  bonds      (2)  $3\pi$  and  $7\sigma$  bonds      (3)  $2\pi$  and  $7\sigma$  bonds      (4)  $4\pi$  and  $8\sigma$  bonds
7. In  $(\text{CH}_3)_4\text{C}$  how many type of degree of carbon present  
 (1) Primary and tertiary      (2) Primary and quaternary      (3) Only primary      (4) Only secondary
8. Which of the following is tertiary chloride ?  
 (1)  $(\text{CH}_3)_2\text{CHCl}$       (2)  $(\text{CH}_3)_3\text{CCH}_2\text{Cl}$       (3)  $(\text{CH}_3)_3\text{CCl}$       (4)  $\text{CH}_3\text{CH}_2\text{Cl}$
9. The homologue of  $\text{CH}_3 - \text{CH}_2 - \text{CH}_3$  is :  
 (1)  $\text{CH}_2 = \text{CH} - \text{CH}_3$       (2)  $\text{CH}_3 - \text{CH}_3$       (3)       (4)  $\text{CH}_3 - \text{C} \equiv \text{CH}$
10. The general formula of  $\text{C}_4\text{H}_4$  is :  
 (1)  $\text{C}_n\text{H}_{2n}$       (2)  $\text{C}_n\text{H}_{2n+2}$       (3)  $\text{C}_n\text{H}_{2n-4}$       (4)  $\text{C}_n\text{H}_{2n-2}$

