



TARGET : NEET (UG) 2024

Course : SARANSH (Youtube Live CRASH COURSE)

PHYSICS

**DPP**

DAILY PRACTICE PROBLEMS

**DPP NO. 1**

**PHYSICS: MODERN PHYSICS**

**DPP No. : 1**

- An electron of mass 'm', when accelerated through a potential V has de-Broglie wavelength  $\lambda$ . The de-Broglie wavelength associated with a proton of mass M accelerated through the same potential difference will be:
 

(1)  $\lambda \sqrt{\frac{M}{m}}$       (2)  $\lambda \sqrt{\frac{m}{M}}$       (3)  $\lambda \left(\frac{M}{m}\right)$       (4)  $\lambda \left(\frac{m}{M}\right)$
- Two hydrogen atoms are in excited state with electrons residing in  $n = 2$ . First one is moving towards left and emits a photon of energy  $E_1$  towards right. Second one is moving towards left with same speed and emits a photon of energy  $E_2$  towards left. Taking recoil of nucleus into account during emission process
 

(1)  $E_1 > E_2$       (2)  $E_1 < E_2$       (3)  $E_1 = E_2$       (4) information insufficient
- Consider atoms H,  $\text{He}^+$ ,  $\text{Li}^{++}$  in their ground states. If  $L_1$ ,  $L_2$  and  $L_3$  are magnitude of angular momentum of their electrons about the nucleus respectively then :
 

(1)  $L_1 = L_2 = L_3$       (2)  $L_1 > L_2 > L_3$       (3)  $L_1 < L_2 < L_3$       (4)  $L_1 = L_2 = L_3$
- The voltage applied to an X-ray tube is 18 kV. The maximum mass of photon emitted by the X-ray tube will be:
 

(1)  $2 \times 10^{-13}$  kg      (2)  $3.2 \times 10^{-36}$  kg      (3)  $3.2 \times 10^{-32}$  kg      (4)  $9.1 \times 10^{-31}$  kg
- When a metallic surface is illuminated with monochromatic light of wavelength  $\lambda$ , the stopping potential is  $5V_0$ . When the same surface is illuminated with light of wavelength  $3\lambda$ , the stopping potential is  $V_0$ . Then the work function of the metallic surface is :
 

(1)  $\frac{hc}{6\lambda}$       (2)  $\frac{hc}{5\lambda}$       (3)  $\frac{hc}{4\lambda}$       (4)  $\frac{2hc}{4\lambda}$
- The energy difference between the first two levels of hydrogen atom is 10.2 eV. What is the corresponding energy difference for a singly ionized helium atom ?
 

(1) 10.2 eV      (2) 20.4 eV      (3) 40.8 eV      (4) 81.6 eV



7. If the binding energy of the electron in a hydrogen atom is 13.6 eV, the energy required to remove the electron from the first excited state of  $\text{Li}^{2+}$  is :
- (1) 30.6 eV                      (2) 13.6 eV                      (3) 13.6 eV                      (4) 122.4 eV
8. Consider a hypothetical annihilation of a stationary electron with a stationary positron. What is the wavelength of resulting radiation.
- (1)  $\frac{h}{2m_0c}$                       (2)  $\frac{h}{m_0c}$                       (3)  $\frac{2h}{m_0c}$                       (4)  $\frac{h}{m_0c^2}$
- (h = Plank's constant, c = speed of light,  $m_0$  = rest mass)
9. Photoelectric effect can be explained by assuming that light
- (1) is a form of transverse waves                      (2) is a form of longitudinal waves  
(3) can be polarised                      (4) consists of quanta
10. The ratio of deBroglie wavelengths of a proton and an alpha particle of same energy is .
- (1) 1                      (2) 2                      (3) 4                      (4) 0.25

